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Structural Analysis of NaYF4 Solution Processed Nanoparticles for Hela Cell Studies

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ABSTRACT

The NaYF₄ nanoparticles were prepared and analyzed. Its structural analysis confirmed the formation of nanocrystals of desired sizes and spectral properties that can be incorporated into Hela cell studies. Internalization of NaYF₄ nanoparticles in HeLa cells were determined at different nanoparticles concentrations and for incubation periods from 3 to 24 hours using various techniques. The images revealed a redistribution of nanoparticles inside the cell that increased with incubation time, concentration levels and depended on the presence of the transfection factor. The study identified factors responsible for an effective endocytosis of the NaYF₄ nanoparticles to HeLa cells. Thus this procedure or method could be applied to investigate a wide range of future "smart" theranostic agents that may result into be very promising fluorescent probes for imaging real-time cellular dynamics.

Keywords: lanthanide, nanocrystal, polyvinylpyrrolidone, HeLa cell, incubation, endocytosis



INTRODUCTION

Rapid advances of nanotechnology have led to the development of new unique nanoparticles that emit light after excitation with NIR energy. Reducing nanoparticles size has stimulated its inclusion into enhanced cellular penetration. This allowed intracellular imaging applications in photodynamic therapy (Manikandan et al., 2014). Particles in the Quantum dots size (Hill & Howard, 1987) and metallic nanoparticles levels (Peters et al., 2014) have been extensively studied as two-photon fluorescent biomarkers in living cells and other biological applications. Twophoton emission spectra obtained using NIR excitation indicate that it strongly depends on the shape of the nanocrystals involved and its environment. There is an ongoing need for research on new materials to help in overcoming various problems associated with toxicity. In vivo imaging using a UV visible radiation has shown several disadvantages. It has a small penetration depth into the tissue, and the possibility of damaging the sample under test (Leite et al., 2002). Therefore, Biophysicist has proposed the "single photon" fluorescent that emits a low energy photon as a result of a higher energy photon absorption in UV range. The use of femtosecond pulses permits to reduce the thermal load causing tissue damage (Walter et al., 2014; Karpov et al., 2014). Using this technology, its sensitivity and resolution can be improved when "multiphoton" process and near-infrared light (NIR) is applied to excite the samples under test. Therefore, the use of near-infrared (NIR) radiation within the "optical transmission window" (780 - 1200 nm) of biological tissues has many advantages. These can be listed as an increase of optical contrast, a greater depth of light penetration, minimized auto-fluorescence and even a reduced light scattering (Hill & Howard, 1987). Its established that, NIR radiation does not cause damage to cellular functions and structures (Manikandan et al., 2014) and thus by reducing the damage will allows bio-dynamical imagining. This is a technique that requires long duration of exposure to the (NIR) irradiation hence showing efficient and stable optical conversion of multiphoton fluorescence,

after excitation by femtosecond NIR. These are suitable for imaging of the time-dependent intracellular processes (Kumar Srivastav et al., 2013). Several literature reports have described NaYF₄ nanocrystals of various shapes and sizes (Leite et al., 2002). An important factor is the choice of the host material which should have the crystal structure matched to ion-doped and low phonon energy to minimize non-radiative relaxation processes and to maximize radiative emission. It's proposed that NaYF₄ crystals can satisfy those mentioned criterion (Manikandan et al., 2014). Therefore, this paper reports the investigation on the concentration of the absorber (Yb³⁺ dopant ions) in order to maximize efficiency.

MATERIALS AND METHODS

Growth and Synthesis of NaYF4 nanoparticles

All reagents used were bought from Sigma-Aldrich. To obtain NaYF⁴ various amounts of Er^{3+} , Yb³⁺ nanoparticles were synthesized whereby Y₂O₃ (1-x mmol), Yb₂O₃ (x mmol, where x = 0.1; 0.18; 0.2; 0.3 mmol and 0.02 mmol) was dissolved in 10% HNO₃ (10 ml). After complete reaction, it was heated to evaporate water. Thereafter, ethylene glycol (10 ml) was added to dissolve the obtained rare Y- earth nitrate. Polyvinylpyrrolidone - PVP40 (0.556 g) and NaCl (1 mmol) was added and the solution was then heated to 80 °C until a homogeneous solution. Finally, NH₄F (4 mmol) was dissolved in ethylene glycol (10 ml) at 80 °C and then added drop wise to a solution of nitrate.

Preparation of NaYF4 into HeLa scientific cells

HeLa is a <u>cell</u> type in an <u>immortal cell line</u> used in scientific research. In this study, a sample of the HeLa cells was bought from Sigma Aldrich suppliers. Cells were cultured in 6 well plates dishes (6 x 10 cm²) at a density of 100 000 / plate. Cultures of cells were kept at 37 °C in a humidified atmosphere containing 5% CO₂. 10 µl of sonicated initial colloidal solution of NaYF₄ in water was dissolved in 500 µl H₂O mixed and incubated for 20 minutes. Lipofectamine 2000 was used to introduce NaYF₄ nanoparticles into the cells. 20µl Lipofectamine 2000 was dissolved in 500µl miliR H₂O and incubated for 15 minutes. They were then incubated together. 500 µl of this solution was added to 10 cm² dish with HeLa cells and incubated at different times. Then different sets of variables were tested the incubation times of 3 h, 6 h, 12 h and 24 h; concentration of nanoparticles: 0.2 µM, 0.4 µM, 2.0 µM, 4.0 µM, 20 µM and 40 µM respectively.

Characterization of NaYF₄ nanoparticles

The crystal structure was determined using the Philips X'Pert Pro Alpha1 MPD diffractometer in the range of 20 (10° - 150°) for 15 hours while the size of the crystals was determined a full width half maximum (FWHM) using the Scherrer formula by using Fullprof. TEM measurements were performed in methanol dispersion while surface morphology was analyzed by Auriga - Zeiss field emission scanning electron microscopy (FE-SEM). Elemental analysis was done using QUANTAX 400 Bruker Energy-dispersive X-ray spectroscopy system while the fluorescence spectrum was measured with a 960 nm continues wave (CW) laser diode with 194 mW power and CCD detection (Spex 270M).

Characterization treated HeLa scientific cells

The HeLa cells were analyzed using a Zeiss 710 NLO fluorescence confocal microscope with a NIR femtosecond laser by observing two channels: one with excitation 980 nm (femtosecond laser) and emission range from 515 to 671 nm; the second with excitation 488 nm (continues wave CW) and emission range from 508 to 585 nm and the data analyzed by CellProfiler program.

RESULTS AND DISCUSSION

NaYF₄ nanoparticles were characterized using the Transmission Electron Microscopy (TEM), X-ray diffraction, and photoluminescence and then they were introduced to HeLa cells and luminescence was observed. The impact of incubation time and its distribution inside cells was examined. Lipofectamine 2000 was used a transfection factor was applied in order to improve the efficiency of nanoparticles uptake into cells. The distribution of NaYF₄ nanoparticles, as a function of incubation time and concentration in the presence and absence of the cellular transporter, was analyzed inside HeLa cells.

Structural characterization of NaYF4 nanoparticles

X-ray diffraction data obtained were consistent with cubic structured NaYF₄nanocrystals confirming that the literature on obtaining similar mixtures of cubic and hexagonal NaYF₄nanocrystals (Manikandan et al., 2014). It was concluded that the method and conditions used in this study were accurate. The size of the nanocrystals evaluated from the Scherrer formula (Gavrilović et al., 2014) ranged from 30 nm for nanoparticles.



Figure 1: X-ray diffraction pattern of NaYF4 nanoparticles

NaYF₄ lattice constant was obtained from calculations using the data from the XRD (Basith et al., 2014). Its lattice constants were calculated using the Rietveld method. As a result NaYF₄ network the lattice parameters changed from 5.52 Å to 5.51 Å as illustrated in figure 2. However, no visible change of lattice constant depending on the content of ytterbium was observed. The lattice constant obtained for NaYF₄ nanoparticles without dopant was 5.518 and this was greater than that of bulk crystal as published in the JCPDS card (No. 77-2042, a = 5.470 Å). Similar results are reported in stresses in lattice in the nanocrystals than in bulk materials (Karpov et al., 2014).



Figure 2. NaYF4 lattice constant obtained from calculations

Elemental EDX characterization of NaYF4 nanoparticles

Elemental EDX analysis of NaYF₄ nanoparticles was performed for all of the nanoparticles samples and obtained as: ions (Yb³⁺) to ions (Er³⁺) was about 11.2 ratio $F^{3+}/Y^{3+} \sim 5.6 Y^{3+}/Er^{3+} \sim 31.6 Y^{3+}/Yb^{3+} \sim 2.7 Y^{3+}/Na^{3+} \sim 1.3; F^{3+}/Na^{3+} \sim 6.2$ respectively. The results showed that the reaction ratios agree with the ratios obtained from the quantitative EDX analysis (Ebrahimi et al., 2014; Walter et al., 2014).

TEM characterization of NaYF₄ nanoparticles

TEM images of NaYF⁴nanoparticles showed a well-defined crystal structure as depicted in figure 4 that was polyhedral in shape and uniform in size with average size around 30 nm. The apparent size differences were statistically insignificant based on 100-160 individual nanostructures per sample TEM determinations. Black areas spotted in the figure 4 were attributed to burned spots during the electron beam exposure during TEM measurements.





Figure 4: Transmission electron microscope image

Luminescence characterization of NaYF4 nanoparticles

Only ions incorporated in the crystal lattice were analyzed using Rietveld method based on X-ray diffraction measurements. Rietveld refinements showed that the Y^{3+} site was partially occupied by Yb^{3+} ions. All Er^{3+} ions occupied the Y^{3+} sites (Leite et al., 2002; Kumar Srivastav et al., 2013). These calculations provided Yb^{3+} fractional occupancy lower than the technological fraction from pure NaYF₄ nanoparticles A linear dependence of luminescence intensity versus Yb^{3+} occupancy obtained and illustrated figure 3 supports the expectation that only site-located Yb^{3+} ions contribute to luminescence.





Fluorescence spectra showed emission peaks at 522, 541 and 652 nm are the result from ${}^{4}H_{11/2}$, ${}^{4}S_{3/2}$, ${}^{4}F_{9/2}$ to ${}^{4}I_{15/2}$ transition of Er^{3+} ions, respectively (Leite et al., 2002). The bands did not depend on the size of nanoparticles but were found to can change the intensity of the fluorescence peaks of the ions. For nanoparticles NaYF₄ strong red fluorescence was observed (Kumar Srivastav et al., 2013) where the ratio of red to green luminescence changes depend on the amount of ions of ytterbium in lattice crystal. This agrees with the amount of ions in a lattice crystal designated from the Rietveld method (Ebrahimi et al., 2014; Di Giovanni, 2014). Fluorescent bands showed a well-defined structure as opposed to the fluorescent bands generated by quantum dots (Peters et al., 2014) or gold nanoparticles. This was attributed to simultaneous emission of the different close electron energy levels of Er^{3+} ions. The emission bands were narrower than the quantum dots (Hameed et al., 2014) or gold nanoparticles (Young et al., 1993) resulting in higher contrast images through better spectral distinction from the auto fluorescence of cells. These advantages make NaYF₄ nanoparticles excellent candidates for the next generation of nanoprobes for *in vivo* bioluminescence imaging.



Figure 6: Fluorescence spectra of NaYF₄

HeLa cells with NaYF4 nanoparticles

Aqueous suspensions of NaYF₄ nanoparticles were introduced into the HeLa cancer cells by simple incubation to obtain high-contrast two-photon fluorescence (Grujić-Brojčin et al., 2014). The results demonstrated that for short incubation times, NaYF₄ nanoparticles were homogeneously distributed inside the cell (Young et al., 1993).while for longer incubation times, the spatial distribution of NaYF₄ nanoparticles were highly inhomogeneous (Manikandan et al., 2014). Images of single cells were obtained after different incubation times whereby confocal images of HeLa cells after 3 h, 6 h, 12 h and 24 h incubation showed that prolonged incubation time, increased quantities of nanoparticles inside the HeLa cells (figure 8). On the other hand, for longer incubation times (24 h with and 24 h without the vesicles) nanoparticles were mainly localized in the vicinity of the nucleus. This was attributed to the spatial redistribution of the nanoparticles is caused by the transport of nutrients from the cell membrane to the vicinity of the cell nucleus (Riwotzki et al., 2001).



Figure 7: Confocal images of HeLa cells after 3 h, 6 h, 12 h and 24 h incubation times.



Figure 8: The ratios indifferent incubation time as analyzed by CellProfiler program

It has been observed that when the concentration of nanoparticles increases by 10 times, some nanoparticles appear in the cytoplasm inside cells. In the case of nanoparticles incubated with cells in the presence of Lipofectamine 2000 much more nanoparticles enter into cells (Maiyalagan et al., 2014) in comparison to incubation without Lipofectamine 2000.

CONCLUSIONS

Small, pure cubic phase NaYF⁴ nanoparticles were synthesized using PVP as a chelating agent. These nanoparticles were uniform and their size varied depending on the experimental conditions. The crystals were monodisperse in water and images of single cells obtained after different incubation times and with or without the liposomes showed the varied dynamic behavior inside the cell. Incubation with the liposomes causes enhanced incorporation of nanoparticles. This result presents an evidence of potential applications of NaYF⁴ nanoparticles to static and dynamic bioimaging. Similar observations were made based on the nanoparticles concentration dependent study in the presence and absence of the transporter. It was then concluded that a good selection of nanoparticles concentration and time of incubation has critical influence on the fabrication and development of multimodal imagining materials.

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